

Polymer and Colloid Highlights

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Interaction of Rare-earth-based Nanoparticles with Carbon Dioxide

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Increasing carbon dioxide production has become a challenge for our society. Nowadays, scientists and engineers are intensifying efforts to develop solutions to mitigate climate change. The strategies can entail many things, for example developing new materials for capture, transport, storage, but also for an efficient conversion of CO₂ to useful chemicals. Carbon dioxide detection is an indispensable part of any of these approaches, but the operating conditions can be extremely different.

Today, CO₂-sensing is still performed with infrared spectroscopy owing to trade-offs between the technological and economical factors and having limited, or no portability. Chemo-resistive sensing is an attractive alternative. However, among different inorganic materials only rare-earth-based sensors, particularly lanthanum oxycarbonate-based, show significant resistance changes upon exposure to CO₂ at elevated

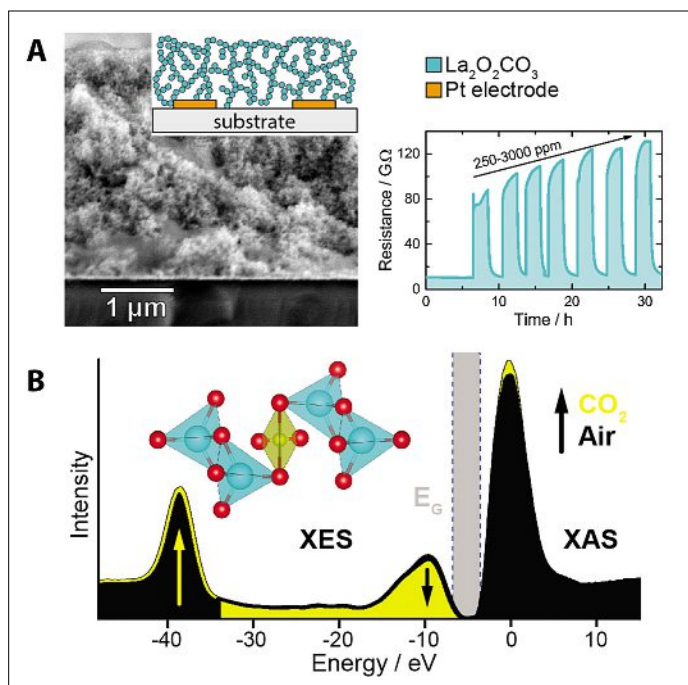


Fig. 1. A: SEM image and scheme of La₂O₂CO₃-film and DC resistance changes during exposure to CO₂ pulses. B: *in situ* HERFD-XAS and vtc-XES spectra changes upon exposure to 1% CO₂ at 250 °C.

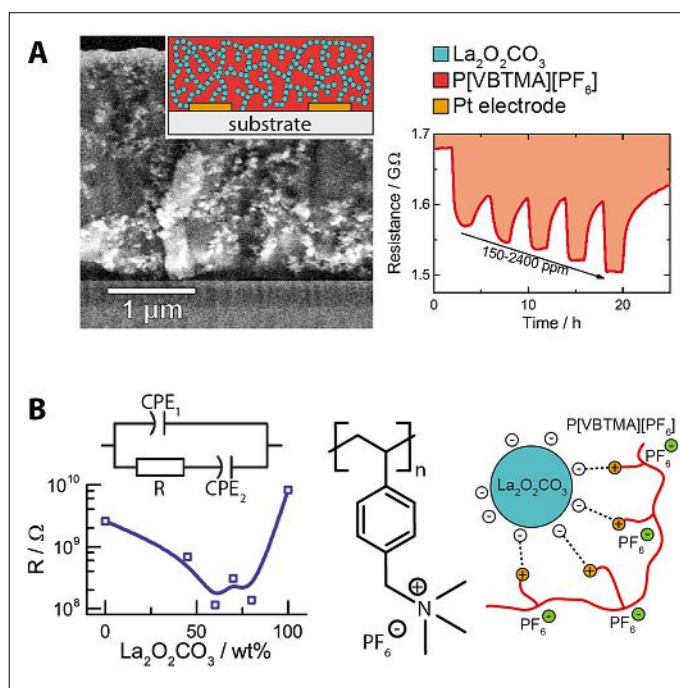


Fig. 2. A: SEM image and scheme of composite film, DC resistance changes during exposure to CO₂ pulses at RT. B: Model circuit of composite film with evolution of baseline resistance as a function of inorganic particle content (left). Model of P[VBTMA][PF₆]/La₂O₂CO₃ interface (right).

temperatures (Fig. 1A). To reveal the uniqueness of La₂O₂CO₃ particles, we studied their atomic and electronic structure by means of X-ray diffraction, high-energy resolution fluorescence detected X-ray absorption (HERFD XAS) and valence-to-core X-ray emission spectroscopy (vtc-XES). The *in situ* studies show that upon interaction with relatively inert CO₂ the distribution of unoccupied lanthanum d-states and occupied O p- and La d-states change significantly (Fig. 1B).^[1] Nevertheless, the realization of a low-power, room temperature chemo-resistive-sensor cannot be achieved based solely on La₂O₂CO₃. Thus, we utilize poly[(p-vinylbenzyl) trimethylammonium hexafluorophosphate], P[VBTMA][PF₆], as a matrix for La₂O₂CO₃ (Fig. 2A). We are taking advantage of the interaction at the organic/inorganic interface to boost the overall conductivity of composites at room temperature (Fig. 2B), which brings realization of chemo-resistive CO₂ sensing within reach.^[2]

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